COURSE OUTLINE

1. **Course:** CHEM 213, Foundations of Chemistry: Change and Equilibrium - Winter 2019

   **Lecture 01:** TR 11:00 - 12:15 in ICT 122

   **Instructor**        **Email**       **Phone**       **Office**       **Hours**
   Julie Lefebvre        jlefebv@ucalgary.ca  403 220-7602  EEEL 237C  TBA

   Lab Coordinator: Dr. Julie Lefebvre

   Start date of Labs: January 21st, 2019

   **Course Site:**

   D2L: CHEM 213 L01 - (Winter 2019) - Foundations of Chemistry: Change and Equilibrium

   **Note:** Students must use their U of C account for all course correspondence.

2. **Requisites:**

   See section 3.5.C in the Faculty of Science section of the online Calendar.

   **Prerequisite(s):**
   Chemistry 201 or 211.

   **Antirequisite(s):**
   Credit for Chemistry 213 and any of 203, 209 or 301 will not be allowed.

   **Note(s):**
   a. Strongly recommended for students majoring in Chemistry and other students with strong backgrounds in chemistry.

3. **Grading:**

   The University policy on grading and related matters is described in F.1 and F.2 of the online University Calendar. In determining the overall grade in the course the following weights will be used:

<table>
<thead>
<tr>
<th>Component(s)</th>
<th>Weighting (%)</th>
<th>Important Dates</th>
</tr>
</thead>
<tbody>
<tr>
<td>Laboratory - Wet Experiments (5)</td>
<td>20</td>
<td>Starts Jan. 28th</td>
</tr>
<tr>
<td>Laboratory - CTDs (5)</td>
<td>15</td>
<td>Starts Jan. 21st</td>
</tr>
<tr>
<td>Graded In-Class Activities (3)</td>
<td>12</td>
<td>Jan. 29th (2%), Feb. 14th (4%), March 28th (6%)</td>
</tr>
<tr>
<td>Midterm</td>
<td>18</td>
<td>March 7th, In class</td>
</tr>
<tr>
<td>Final Examination</td>
<td>35</td>
<td></td>
</tr>
</tbody>
</table>

   Each piece of work (reports, assignments, quizzes, midterm exam(s) or final examination) submitted by the student will be assigned a grade. The student's grade for each component listed above will be combined with the indicated weights to produce an overall percentage for the course, which will be used to determine the course letter grade.
The conversion between a percentage grade and a letter grade is as follows.

<table>
<thead>
<tr>
<th>Minimum % Required</th>
<th>A+</th>
<th>A</th>
<th>A-</th>
<th>B+</th>
<th>B</th>
<th>B-</th>
<th>C+</th>
<th>C</th>
<th>C-</th>
<th>D+</th>
<th>D</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>92.00 %</td>
<td>86.00 %</td>
<td>81.00 %</td>
<td>76.00 %</td>
<td>72.00 %</td>
<td>68.00 %</td>
<td>62.00 %</td>
<td>58.00 %</td>
<td>54.00 %</td>
<td>50.00 %</td>
<td>45.00 %</td>
</tr>
</tbody>
</table>

This course has a registrar scheduled final exam.

Notes:

Students will be expected to understand at every stage the material covered in all components of the course. In order to satisfy the prerequisite requirements (i.e., C-) for further Chemistry courses, a student must meet the following requirements: (1) achieve a minimum 50% in the laboratory grading, (2) complete no less than seven (out of the ten) laboratory sessions (including submission of worksheets or report for each session), and (3) achieve either a minimum 50% on the final examination, or a minimum 50% weighted average on the examinations (Midterm and Final).

This means that if a student scores below 50% in either the laboratory component or the examinations, or fails to complete seven laboratory sessions, then the maximum course letter grade they can obtain in CHEM 213 is a D+.

4. Missed Components Of Term Work:

In the event that a student misses the midterm or any course work due to illness, supporting documentation, such as a medical note or a statutory declaration will be required (see Section N.1; for more information regarding the use of statutory declaration/medical notes, see FAQ). Absences must be reported within 48 hrs.

The regulations of the Faculty of Science pertaining to this matter are found in the Faculty of Science area of the Calendar in Section 3.6. It is the student’s responsibility to familiarize themselves with these regulations. See also Section E.3 of the University Calendar.

There are no deferred Midterm/term test examinations. In the event that a student misses the midterm or any course work due to illness then an official medical note will be required. Absences must be reported within 48 hrs. If a student misses the midterm for other reasons, then analogous documentation will be required. The course coordinator will need to see the original documentation (not electronic copy) for review / decision and keep it (or a copy) for their records. The documentation must be provided to the course coordinator within 15 days of the date of the midterm in order for an excused absence to be considered. If an excused absence is approved, then the percentage weight of a legitimately missed midterm examination will be transferred to the final examination.

If a student missed an experiment or a make-up lab for non-legitimate reasons (e.g. vacation, incomplete or insufficient score in pre-lab assignment), and did not perform the experiment, the contribution of that experiment in the final course grade will be zero.

Make-up lab policy for Chem 213: There are no formal make-up labs in CHEM 213. In the event that a student misses a laboratory activity for a legitimate reason, this activity will be rescheduled (if possible) later during the same week or an excused absence will be granted and the weight of the legitimately missed laboratory activity will be pro-rated among the other laboratory activities.

5. Scheduled Out-of-Class Activities:

There are no scheduled out-of-class activities for this course.
6. **Course Materials:**

   Recommended Textbook(s):
   
   Mahaffy, Bucat, Tasker et al., *Chemical - Human Activity, Chemical Reactivity*: Nelson.

   **Additional Material:**
   
   - A molecular model kit
   - A self-duplicating laboratory notebook
   - A full length lab coat
   - Safety glasses
   - A lock for lab drawer (combination preferred)
   - A non-programmable scientific calculator (such as Casio FX 260)

   **Online Course Components:** Students may opt to participate in lecture question activities using TopHat. This will be described in detail on the first day of class. Though this is not mandatory, students are strongly recommended to participate to build up concepts and problem solving skills. The average mark obtained on TopHat could be used to replace one low (non-zero) CTD mark.

7. **Examination Policy:**

   During exams students are allowed to bring only pencils, pens, erasers, their ID card, a molecular model kit and non-programmable calculators (recommended calculator is the CASIO fx-260 solar). If in doubt, check your calculator with your instructor prior to the examination (the programmable TI calculators are not acceptable). Students should also read the Calendar, **Section G**, on Examinations.

   Special Needs Students must be registered with Student Accessibility Services (see section 11(c)), and must identify themselves to their instructor as soon as possible.

   Students should also read the Calendar, **Section G**, on Examinations.

8. **Approved Mandatory And Optional Course Supplemental Fees:**

   **Laboratory Breakage Fees and Locker Check-out:** The Department of Chemistry has a laboratory glassware breakage fee. At the start of the course, each student is assigned a locker and checks-in to establish that they have a complete set of usable glassware. By signing for check-in, a student agrees that they are now responsible for the glassware until check out. Any equipment that is missing, unusable or has been replaced during the semester will be charged to the student. All students, even those who withdraw early from the course must check out of the laboratory before the last day of lectures [April 12th, 2019]. Any student who fails to check out before the last day of lectures for the term will be assessed a charge of $30.00. If this fee is not paid by the last day of the payment period (Jan 31 for Fall courses, April 30 for Winter courses, July 15 for Spring courses), an additional $10.00 administrative fee will be charged and university services (registration, transcripts, etc.) may be withheld.

9. **Writing Across The Curriculum Statement:**

   For all components of the course, in any written work, the quality of the student's writing (language, spelling, grammar, presentation etc.) can be a factor in the evaluation of the work. See also Section E.2 of the University Calendar.

10. **Human Studies Statement:**

    If you agree, your course work may be used for research purposes. Your responses will remain anonymous and confidential. Grouped data (no individual responses) may be used in academic presentations and publications. Participation in such research is voluntary and will not influence grades in this course. Students' signed consent forms will be withheld from instructors until after final grades are submitted. More information will be provided at the time student participation is requested.

    See also **Section E.5** of the University Calendar.

11. **Reappraisal Of Grades:**
A student wishing a reappraisal, should first attempt to review the graded work with the Course coordinator/instructor or department offering the course. Students with sufficient academic grounds may request a reappraisal. Non-academic grounds are not relevant for grade reappraisals. Students should be aware that the grade being reappraised may be raised, lowered or remain the same. See Section I.3 of the University Calendar.

a. **Term Work:** The student should present their rationale as effectively and as fully as possible to the Course coordinator/instructor within 15 days of either being notified about the mark, or of the item's return to the class. If the student is not satisfied with the outcome, the student shall immediately submit the Reappraisal of Graded Term work form to the department in which the course is offered. The department will arrange for a re-assessment of the work if, and only if, the student has sufficient academic grounds. See sections I.1 and I.2 of the University Calendar.

b. **Final Exam:** The student shall submit the request to Enrolment Services. See Section I.3 of the University Calendar.

12. **Other Important Information For Students:**

a. **Mental Health** The University of Calgary recognizes the pivotal role that student mental health plays in physical health, social connectedness and academic success, and aspires to create a caring and supportive campus community where individuals can freely talk about mental health and receive supports when needed. We encourage you to explore the mental health resources available throughout the university community, such as counselling, self-help resources, peer support or skills-building available through the SU Wellness Centre (Room 370, MacEwan Student Centre, Mental Health Services Website) and the Campus Mental Health Strategy website (Mental Health).

b. **SU Wellness Center:** The Students Union Wellness Centre provides health and wellness support for students including information and counselling on physical health, mental health and nutrition. For more information, see www.ucalgary.ca/wellnesscentre or call 403-210-9355.

c. **Sexual Violence:** The University of Calgary is committed to fostering a safe, productive learning environment. The Sexual Violence Policy (https://www.ucalgary.ca/policies/files/policies/sexual-violence-policy.pdf) is a fundamental element in creating and sustaining a safer campus environment for all community members. We understand that sexual violence can undermine students' academic success and we encourage students who have experienced some form of sexual misconduct to talk to someone about their experience, so they can get the support they need. The Sexual Violence Support Advocate, Carla Bertsch, can provide confidential support and information regarding sexual violence to all members of the university community. Carla can be reached by email (svsa@ucalgary.ca) or phone at 403-220-2208.

d. **Misconduct:** Academic misconduct (cheating, plagiarism, or any other form) is a very serious offence that will be dealt with rigorously in all cases. A single offence may lead to disciplinary probation or suspension or expulsion. The Faculty of Science follows a zero tolerance policy regarding dishonesty. Please read the sections of the University Calendar under Section K. Student Misconduct to inform yourself of definitions, processes and penalties. Examples of academic misconduct may include: submitting or presenting work as if it were the student's own work when it is not; submitting or presenting work in one course which has also been submitted in another course without the instructor's permission; collaborating in whole or in part without prior agreement of the instructor; borrowing experimental values from others without the instructor's approval; falsification/fabrication of experimental values in a report. These are only examples.

e. **Assembly Points:** In case of emergency during class time, be sure to FAMILIARIZE YOURSELF with the information on assembly points.

f. **Academic Accommodation Policy:** Students needing an accommodation because of a disability or medical condition should contact Student Accessibility Services in accordance with the procedure for accommodations for students with disabilities available at procedure-for-accommodations-for-students-with-disabilities.pdf.

Students needing an accommodation in relation to their coursework or to fulfill requirements for a graduate degree, based on a protected ground other than disability, should communicate this need, preferably in writing, to the Associate Head of the Department of Chemistry, Dr. Farideh Jalilehvand by email ahugchem@ucalgary.ca or phone 403-220-5353. Religious accommodation requests relating to class, test or exam scheduling or absences must be submitted no later than 14 days prior to the date in question. See Section E.4 of the University Calendar.
g. **Safewalk:** Campus Security will escort individuals day or night (See the [Campus Safewalk](#) website). Call 403-220-5333 for assistance. Use any campus phone, emergency phone or the yellow phones located at most parking lot pay booths.

h. **Freedom of Information and Privacy:** This course is conducted in accordance with the Freedom of Information and Protection of Privacy Act (FOIPP). Students should identify themselves on all written work by placing their name on the front page and their ID number on each subsequent page. For more information, see [Legal Services](#) website.

i. **Student Union Information:** VP Academic, Phone: 403-220-3911 Email: suvpaca@ucalgary.ca. SU Faculty Rep., Phone: 403-220-3913 Email: scincerep@su.ucalgary.ca. Student Ombudsman, Email: suvpaca@ucalgary.ca.

j. **Internet and Electronic Device Information:** Unless instructed otherwise, cell phones should be turned off during class. All communication with other individuals via laptop, tablet, smart phone or other device is prohibited during class unless specifically permitted by the instructor. Students that violate this policy may be asked to leave the classroom. Repeated violations may result in a charge of misconduct.

k. **Surveys:** At the University of Calgary, feedback through the Universal Student Ratings of Instruction (USRI) survey and the Faculty of Science Teaching Feedback form provides valuable information to help with evaluating instruction, enhancing learning and teaching, and selecting courses. Your responses make a difference - please participate in these surveys.

l. **Copyright of Course Materials:** All course materials (including those posted on the course D2L site, a course website, or used in any teaching activity such as (but not limited to) examinations, quizzes, assignments, laboratory manuals, lecture slides or lecture materials and other course notes) are protected by law. These materials are for the sole use of students registered in this course and must not be redistributed. Sharing these materials with anyone else would be a breach of the terms and conditions governing student access to D2L, as well as a violation of the copyright in these materials, and may be pursued as a case of student academic or non-academic misconduct, in addition to any other remedies available at law.

13. **Laboratory Information**

Laboratory activities will begin on January 21st. It is mandatory that students wear a lab coat and safety glasses at all times when working in the lab. Students wearing inappropriate laboratory attire will not be permitted to conduct experiments for safety reasons. The manual and additional resources can be found online (course D2L site). You must consult the course website prior to attending any of your scheduled lab periods, complete (if applicable) the pre-lab assignment, and printout (if applicable) the required material that outlines the procedures you will be doing. You must bring your self-duplicating notebook to every lab activity.

Students repeating the course within the last two years can be exempted from the Laboratory Component of the Course if a grade of 75% or higher was obtained. The lab grade achieved on the previous attempt will be carried forward. Such students must contact the Chemistry Undergraduate Program Administrator in the Chemistry Main Office, SA 229 before the drop date (January 17th, 2019).

14. **Laboratory Safety Course**

All undergraduate students taking chemistry laboratories are required to complete an introductory course (approx. 50 minutes) on laboratory safety. This course is presented in an online format. The Safety Course must be completed before the first laboratory experiment. Students who do not complete the safety lessons will subsequently be denied admission to the laboratories. While it will not count directly to the final grade, the material is considered to be part of the course and is therefore appropriate for inclusion into laboratory pre-labs and exams. Students who have previously completed the Chemistry Safety Course at the University of Calgary in the past five years are NOT required to repeat it.

**Course Outcomes:**

- Explain dynamic chemical equilibria and determine qualitatively and quantitatively how to manipulate them on the macroscopic and molecular level.
Define and manipulate the rate of a chemical reaction and examine the relationship between the macroscopic reaction rate and the reaction mechanism happening at the molecular level.

Determine qualitatively and quantitatively the thermodynamic parameters associated with a chemical reaction to explain its spontaneity and describe how to alter it.

Collect and analyze observations to create and evaluate hypotheses about the relationship between the macroscopic and molecular level.

- Perform basic chemical techniques to study equilibrium, kinetics, acid/base, dissolution, precipitation and redox reactions.